



**UNIVERSITAS NEGERI YOGYAKARTA**  
FACULTY OF MATHEMATICS AND NATURAL SCIENCES  
DEPARTMENT OF MATHEMATICS EDUCATION

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**Bachelor of Science in Mathematics**

**MODULE HANDBOOK**

Module name:	Mathematical Modelling
Module level,if applicable:	Undergraduate
Code:	MAT 6322
Sub-heading,if applicable:	-
Classes,if applicable:	-
Semester:	4 <sup>th</sup>
Module coordinator:	Fitriana Yuli Saptaningtyas,M.Si.
Lecturer(s):	Fitriana Yuli Saptaningtyas,M.Si.
Language:	Bahasa Indonesia
Classification within the curriculum:	Compulsory course
Teaching format / class hours perweek during the semester:	150 minutes lectures and 180 minutes structured activities per week.
Workload:	Total workload is 136 hours per semester which consists of 150 minutes lectures, 180 minutes structured activities, and 180 minutes self-study per week for 16 weeks.
Creditpoints:	3
Prerequisites course(s):	Differential Equations (MAT6314)
Course outcomes:	After taking this course the students have ability to: CO1. Demonstrate collaborative attitude and independence in carrying out individual tasks and group assignments CO2. Mastering the concepts and basics of mathematical modelling process CO3. Use Matlab software to solve a simple mathematical

	<p>modelling of real problem</p> <p>CO4. Make a simple mathematical modelling of real problem</p>																										
Content:	<p>The discussion in this course emphasizes the process of modeling a case. Therefore, the discussion material is divided into three basic topics: terminology, examples of mathematical models and mathematical modeling projects.</p>																										
Study/exam achievements:	<p>CO1: Attitude assessment is carried out at each meeting by observation and / or self-assessment techniques using the assumption that basically every student has a good attitude. The student is given a value of very good or not good attitude if they show it significantly compared to other students in general. The result of attitude assessment is not a component of the final grades, but as one of the requirements to pass the course. Students will pass from this course if at least have a good attitude.</p> <p>The final mark will be weight as follow:</p> <table border="1" data-bbox="649 1113 1445 1459"> <thead> <tr> <th>No</th> <th>CO</th> <th>Assesment Object</th> <th>Assessment Techniques</th> <th>Weight</th> </tr> </thead> <tbody> <tr> <td rowspan="4">1</td> <td rowspan="4">CO2 and CO 3</td> <td>a. Individual assignments</td> <td rowspan="4">Written test</td> <td>20%</td> </tr> <tr> <td>b. group assignments</td> <td>10%</td> </tr> <tr> <td>c. Mid Exam</td> <td>25%</td> </tr> <tr> <td>d. Final Exam</td> <td>30%</td> </tr> <tr> <td>2</td> <td>CO 4</td> <td>Presentation and Project</td> <td>Observation</td> <td>15%</td> </tr> <tr> <td colspan="4">Total</td> <td>100%</td> </tr> </tbody> </table>	No	CO	Assesment Object	Assessment Techniques	Weight	1	CO2 and CO 3	a. Individual assignments	Written test	20%	b. group assignments	10%	c. Mid Exam	25%	d. Final Exam	30%	2	CO 4	Presentation and Project	Observation	15%	Total				100%
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2	CO 4	Presentation and Project	Observation	15%																							
Total				100%																							
Forms of media:	Board, LCD Projector, Laptop/Computer																										
Literature:	<p>1. Jonathan Simon (2011), <i>Head First Android Development</i> (e-Maki, Daniel P. Thompson, Maynard. Mathematical Models and Applications. 1973. Prentice-Hall. United States of America.</p> <p>2. Huntley, I.D. James, D.J.G. <i>Mathematical Modelling A Source Book of Case Studies</i>. 1990. Oxford University Press. New York</p>																										

	<p>3. Bender, Edward A. An Introduction to Mathematical Modeling. 1978. John Willey &amp; Sons. Canada</p> <p>4. Tuharto. Diktat Kuliah Pengantar Model Matematika. 2001. Jurusan Pendidikan Matematika FMIPA UNY.</p> <p>5. Giordano, Frank R. Weir, Maurice D. A First Course in Mathematical Modelling 3rd Ed. 2003. Thomson Learning. USA</p>
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**PLO and CO mapping**

	PLO1	PLO2	PLO3	PLO4	PLO5	PLO6	PLO7	PLO8	PLO9	PLO10
CO1		✓								
CO2				✓						
CO3					✓					
CO4							✓			